



DR ACADEMY

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NEET EXAMINATION – 2022

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T6

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PHYSICS - SECTION - A

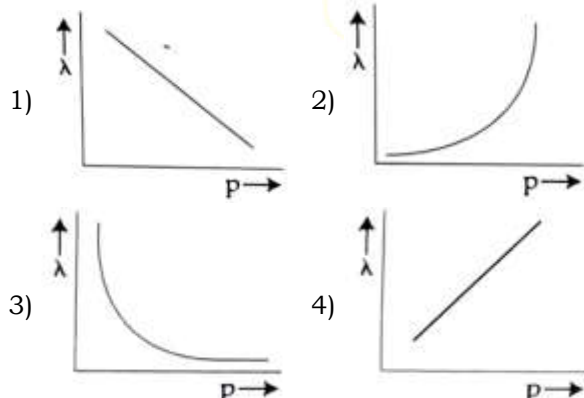
1. If a soap bubble expands, the pressure inside the bubble
- 1) Increases
 - 2) Remain the same
 - 3) Is equal to the atmospheric pressure
 - 4) Decreases

Ans. 4

Sol. $P = \frac{4T}{r}$

If r increases, pressure decreases

2. The graph which shows the variation of the de Broglie wavelength (λ) of a particle and its associated momentum (p) is



Ans. 3

Sol. $\lambda = \frac{h}{p}$

$\lambda \propto \frac{1}{p}$

3. A body of mass 60g experiences a gravitational force of 3.0N, when placed at a particular point. The magnitude of the gravitational field intensity at that point is

- 1) 50 N/kg
- 2) 20 N/kg
- 3) 180 N/kg
- 4) 0.05 N/kg

Ans. 1

Sol. $F = mg$

$E = g = \frac{F}{m} = 50 \text{ N/kg}$

4. Given below are two statements:

Statement-I:- Biot-Savart's law gives us the expression for the magnetic field strength of an infinitesimal current element (Idl) of a current carrying conductor only.

Statement-II: Biot-Savart's law is analogous to Coulomb's inverse square law of charge q , with the former being related to the field produced by a scalar source, Idl while the latter being produced by a vector source, q . In light of above statements choose the most appropriate answer from the options given below

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct and Statement II is incorrect
- 3) Statement I is incorrect and Statement II is correct
- 4) Both statement I and Statement II are correct

Ans. 2

Sol. Conceptual

5. The ratio of the distance travelled by a freely falling body in the 1st, 2nd, 3rd and 4th second
- 1) 1:4:9:16 2) 1:3:5:7
3) 1:1:1:1 4) 1:2:3:4

Ans. 2

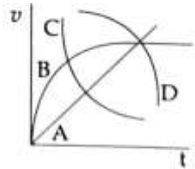
Sol. $S_n = \frac{g(2n-1)}{2}$

6. In half wave rectification, if the input frequency is 60Hz, then the output frequency would be
- 1) 30 Hz 2) 60 Hz 3) 120 Hz 4) Zero

Ans. 2

Sol. Conceptual

7. A spherical ball is dropped in a long column of a highly viscous liquid. The curve in the graph shown, which represents the speed of the ball (v) as a function of time (t) is



- 1) B 2) C 3) D 4) A

Ans. 1

Sol. Conceptual

8. The angle between the electric lines of force and the equipotential surface is
- 1) 45° 2) 90° 3) 180° 4) 0°

Ans. 2

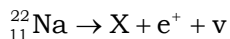
Sol. Conceptual

9. As the temperature increases, the electrical resistance
- 1) Decreases for both conductors and semiconductors
2) Increases for conductors but decreases for semiconductors
3) Decreases for conductors but increase for semiconductors
4) Increase for both conductors and semiconductors

Ans. 2

Sol. Conceptual

10. In the given nuclear reaction, the element X is



- 1) ${}_{10}^{23}\text{Ne}$ 2) ${}_{10}^{22}\text{Ne}$ 3) ${}_{12}^{22}\text{Mg}$ 4) ${}_{11}^{23}\text{Na}$

Ans. 2

Sol. Conceptual

11. Let T_1 and T_2 be the energy of an electron in the first and second excited states of hydrogen atom, respectively. According to the Bohr's model of an atom, the ratio $T_1:T_2$ is
- 1) 4:1 2) 4:9 3) 9:4 4) 1:4

Ans. 3

Sol. $E_n \propto \frac{1}{n^2}$

12. The ratio of the radius of gyration of a thin uniform disc about an axis passing through its centre and normal to its plane to the radius of gyration of the disc about its diameter is
- 1) $\sqrt{2}:1$ 2) 4:1 3) $1:\sqrt{2}$ 4) 2:1

Ans. 1

Sol. $I_c = \frac{MR^2}{2} = MK_c^2 \Rightarrow K_c = \frac{R}{\sqrt{2}}$

$$I_d = \frac{MR^2}{4} = MK_d^2 \Rightarrow K_d = \frac{R}{2}$$

$$\frac{K_c}{K_d} = \frac{\sqrt{2}}{1}$$

13. When light propagates through a material medium of relative permittivity ϵ_r and relative permeability μ_r , the velocity light, v is given by : (c-velocity of light vacuum)

1) $v = \sqrt{\frac{\mu_r}{\epsilon_r}}$ 2) $v = \sqrt{\epsilon_r \mu_r}$

3) $v = \frac{c}{\sqrt{\epsilon_r \mu_r}}$ 4) $v = c$

Ans. 3

Sol. $C = \frac{1}{\sqrt{\mu_0 \epsilon_0}}$

$$V = \frac{1}{\sqrt{\mu \epsilon}}$$

14. The energy that will be ideally radiated by a 100 kW transmitter in 1 hour is

- 1) $36 \times 10^4 \text{ J}$ 2) $36 \times 10^5 \text{ J}$
3) $1 \times 10^5 \text{ J}$ 4) $36 \times 10^7 \text{ J}$

Ans. 4

Sol. $P = \frac{E}{t} \Rightarrow E = Pt$

15. An electric lift with a maximum load of 2000 kg (lift + passengers) is moving up with a constant speed of 1.5 ms^{-1} . The frictional force opposing the motion is 3000N. The minimum power delivered by the motor to the lift in watts is: ($g=10 \text{ ms}^{-2}$)

- 1) 20000 2) 34500
3) 23500 4) 23000

Ans. 2

Sol. $P = F.V$
 $(Mg + f)v$

16. Two resistors of resistance 100Ω and 200Ω are connected in parallel in an electrical circuit. The ratio of the thermal energy developed in 100Ω to that in 200Ω in a given time is

- 1) 2:1 2) 1:4 3) 4:1 4) 1:2

Ans. 1

Sol. $H = \frac{V^2}{R} t \Rightarrow H \propto \frac{1}{R}$
 $\frac{H_1}{H_2} = \frac{R_2}{R_1}$

17. The peak voltage of the ac source is equal to

- 1) The rms value of the ac source
- 2) $\sqrt{2}$ times the rms value of the ac source
- 3) $1/\sqrt{2}$ times the rms value of the ac source
- 4) The value of voltage supplied to the circuit

Ans. 2

Sol. $V_{\text{rms}} = \frac{V_0}{\sqrt{2}}$

18. A shell of mass m is at rest initially. It explodes into three fragments having mass in the ratio 2:2:1. If the fragments having equal mass fly off along mutually perpendicular direction with speed v , the speed of the third (lighter) fragment is

- 1) $\sqrt{2}v$ 2) $2\sqrt{2}v$ 3) $3\sqrt{2}v$ 4) v

Ans. 2

Sol. L.C.M

$$0 = \vec{P}_1 + \vec{P}_2 + \vec{P}_3$$

$$P_3 = \sqrt{P_1^2 + P_2^2}$$

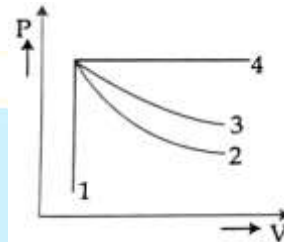
19. If the initial tension on a stretched string is doubled, then the ratio of the initial and final speed of a transverse wave along the string is

- 1) $\sqrt{2} : 1$
- 2) $1 : \sqrt{2}$
- 3) $1 : 2$
- 4) $1 : 1$

Ans. 2

Sol. $V = \sqrt{\frac{T}{\mu}}$
 $\frac{V_1}{V_2} = \sqrt{\frac{T_1}{T_2}}$

20. An ideal gas undergoes four different processes from the same initial state as shown in the figure below. Those processes are adiabatic, isothermal, isobaric and isochoric. The curve which represents the adiabatic process among 1, 2, 3 and 4 is



- 1) 2 2) 3 3) 4 4) 1

Ans. 1

Sol. Conceptual

21. The angular speed of a fly wheel moving with uniform angular acceleration changes from 1200 rpm to 3120 rpm in 16 seconds. The angular acceleration in rad/s^2 is

- 1) 4π 2) 12π 3) 104π 4) 2π

Ans. 1

Sol. $\alpha = \frac{\omega_2 - \omega_1}{t}$

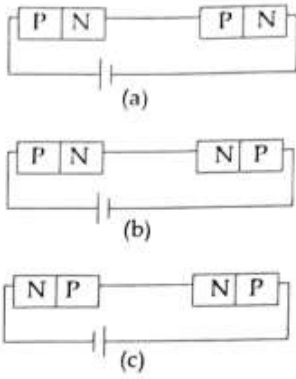
$$\frac{2\pi}{t}(n_2 - n_1)$$

22. Plane angle and solid angle have

- 1) Dimensions but no units
- 2) No units and no dimensions
- 3) Both units and dimensions
- 4) Units but no dimensions

Ans. 4

Sol. Conceptual



23.

In the given circuits (a), (b) and (c), the potential drop across the two p-n junctions are equal in

- 1) Circuit (b) only
- 2) Circuit (c) only
- 3) Both circuits (a) and (c)
- 4) Circuit (a) only

Ans. 3

Sol. Conceptual

24. In a young's double slit experiment, a student observe 8 fringes in a certain segment of screen when a monochromatic light of 600 nm wavelength is used. If the wavelength of light is changed to 400 nm, then the number of fringes he would observe in the same region of the screen is

- 1) 8
- 2) 9
- 3) 12
- 4) 6

Ans. 3

Sol. $n_1\lambda_1 = n_2\lambda_2$

25. When two monochromatic lights of frequency, ν and $\frac{\nu}{2}$ are incident on a photoelectric metal,

their stopping potential becomes $\frac{V_s}{2}$ and V_s respectively. The threshold frequency for this metal is

- 1) 3ν
- 2) $\frac{2}{3}\nu$
- 3) $\frac{3}{2}\nu$
- 4) 2ν

Ans. 3

Sol. $h\nu = h\nu_0 + \frac{eV_s}{2} \dots\dots(1)$

$\frac{h\nu}{2} = h\nu_0 + eV_s \dots\dots(2)$

26. A square loop of side 1 m and resistance 1Ω is placed in a magnetic field 0.5T. If the plane of loop is perpendicular to the direction of magnetic field, the magnetic flux through the loop is

- 1) 0.5 weber
- 2) 1 weber
- 3) Zero weber
- 4) 2 weber

Ans. 1

Sol. $\phi = BA$

27. Two hollow conducting spheres of radii R_1 and R_2 ($R_1 \gg R_2$) have equal charges. The potential would be

- 1) More on smaller spheres
- 2) Equal on both the spheres
- 3) Dependent on the material property of the sphere
- 4) More on bigger sphere

Ans. 1

Sol. $V_1 = \frac{kq}{R_1}$

$V_2 = \frac{kq}{R_2}$

28. A copper wire of length 10 m and radius $(10^{-2} / \sqrt{\pi})$ m has electrical resistance of 10Ω .

The current density in the wire for an electric field strength 10(V/m) is

- 1) 10^6 A/m^2
- 2) 10^{-5} A/m^2
- 3) 10^5 A/m^2
- 4) 10^4 A/m^2

Ans. 3

Sol. $R = \frac{\rho l}{A}$

$J = \sigma E$

$\sigma = \frac{1}{\rho}$

29. A long solenoid of radius 1 mm has 100 turns per mm. If 1A current flows in the solenoid, the magnetic field strength at the centre of the solenoid is

- 1) $12.56 \times 10^{-2} \text{ T}$
- 2) $12.56 \times 10^{-4} \text{ T}$
- 3) $6.28 \times 10^{-4} \text{ T}$
- 4) $6.28 \times 10^{-2} \text{ T}$

Ans. 1

Sol. $B = \mu_0 \cdot n \cdot i$

$= 4\pi \times 10^{-7} \times 100 / 10^{-3} \times 1$

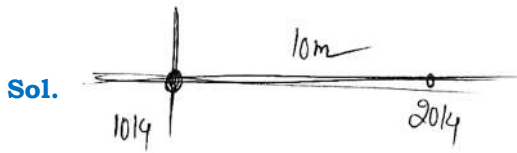
$= 4 \times 3.14 \times 10^{-2}$

$= 12.56 \times 10^{-2}$

30. Two objects of mass 10 kg and 20 kg respectively are connected to the two ends of a rigid rod of length 10m with negligible mass. The distance of the center of mass of the system from the 10kg mass is

- 1) $\frac{20}{3}$ m 2) 10 m 3) 5 m 4) $\frac{10}{3}$ m

Ans. 1



Sol.

$$x_{cm} = \frac{10 \times 10 + 20 \times 10}{30}$$

$$= 200/30$$

$$= 20/3 \text{ m}$$

31. A light ray falls on a glass surface of refractive index $\sqrt{3}$, at an angle 60° . The angle between the refracted and reflected rays would be
1) 60° 2) 90° 3) 120° 4) 30°

Ans. 2

Sol. $\mu = \tan P$

$$\sqrt{3} = \tan 60$$

Angle between reflected and refracted ray is

$$90^\circ$$

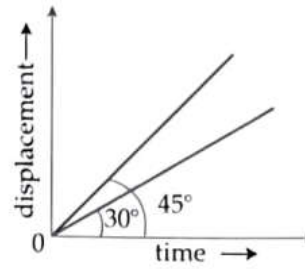
32. The dimensions $[MLT^{-2}A^{-2}]$ belong to the:

- 1) self inductance
2) magnetic permeability
3) electric permittivity
4) magnetic flux

Ans. 2

Sol. $\mu_0 = \frac{F \cdot r^2}{m_1 \cdot m_2} = \frac{MLT^{-2} \cdot L^2}{A^2 \cdot L^2} = MLT^{-2}A^{-2}$

33. The displacement – time graphs of two moving particles make angles of 30° and 45° with the x-axis as shown in the figure. The ratio of their respective velocity is



- 1) 1:1 2) 1:2 3) $1:\sqrt{3}$ 4) $\sqrt{3}:1$

Ans. 3

Sol. Slope = $\tan \theta = \frac{\tan 30}{\tan 45} = \frac{1}{\sqrt{3}}$

34. A biconvex lens has radii of curvature, 20cm each. If the refractive index of the material of the lens is 1.5, the power of the lens is

- 1) +20D 2) +5D 3) infinity 4) +2D

Ans. 2

Sol. $\frac{1}{f} = (\mu - 1) \left(\frac{1}{R_1} - \frac{1}{R_2} \right)$

$$= (0.5) \left(\frac{1}{20} - \frac{1}{-20} \right)$$

$$= (0.5) (2/20)$$

$$= \frac{1}{20}$$

$$P = 5D$$

35. Match List-I with List-II

List-I (Electromagnetic waves)	List-II (Wavelength)
-----------------------------------	-------------------------

- | | |
|-------------------------|-------------------|
| (a) AM radio waves | (i) 10^{-10} m |
| (b) Microwaves | (ii) 10^2 m |
| (c) Infrared radiations | (iii) 10^{-2} m |
| (d) X-rays | (iv) 10^{-4} m |

Choose the correct answer from the options given below.

- 1) (a) - (iii), (b) - (ii), (c) - (i), (d) - (iv)
2) (a) - (iii), (b) - (iv), (c) - (ii), (d) - (i)
3) (a) - (ii), (b) - (iii), (c) - (iv), (d) - (i)
4) (a) - (iv), (b) - (iii), (c) - (ii), (d) - (i)

Ans. 3

Sol. Radio waves- 10^2
Micro waves - 10^{-2} m
I.R - 10^{-4} m
X - 10^{-10} m

PHYSICS - SECTION - B

36. A nucleus mass number 189 splits into two nuclei having mass number 125 and 64. The ratio of radius of two daughter nuclei respectively is

- 1) 4 : 5 2) 5 : 4 3) 25 : 16 4) 1 : 1

Ans. 2

Sol.
$$\frac{R_1}{R_2} = \left(\frac{A_1}{A_2}\right)^{1/3}$$

$$= \left(\frac{125}{64}\right)^{1/3}$$

$$= 5/4$$

37. Two transparent media A and B are separated by a plane boundary. The speed of light in those media are 1.5×10^8 m/s and 2.0×10^8 m/s, respectively. The critical angle for a ray of light for these two media is

- 1) $\sin^{-1}(0.750)$ 2) $\tan^{-1}(0.500)$
 3) $\tan^{-1}(0.750)$ 4) $\sin^{-1}(0.500)$

Ans. 1

Sol.
$$\sin c = \frac{\mu_R}{\mu_D} = \frac{v_D}{v_R} = \frac{1.5 \times 10^8}{2 \times 10^8}$$

$$\sin c = 3/4$$

$$c = \sin^{-1}(0.750)$$

38. A big circular coil of 1000 turns and average radius 10m is rotating about its horizontal diameter at 2 rad s^{-1} . If the vertical component of earth's magnetic field at that place is $2 \times 10^{-5} \text{ T}$ and electrical resistance of the coil is 12.56Ω , then the maximum induced current in the coil will be

- 1) 1.5 A 2) 1 A 3) 2 A 4) 0.25 A

Ans. 2

Sol.
$$i = \frac{BAN\omega}{R}$$

$$i = 1 \text{ A}$$

39. Match List-I with List-II.

List-I	List-II
(a) Gravitational constant (G)	(i) $[L^2T^{-2}]$
(b) Gravitational potential energy	(ii) $[M^{-1}L^3T^{-2}]$
(c) Gravitational potential	(iii) $[LT^{-2}]$
(d) Gravitational intensity	(iv) $[ML^2T^{-2}]$

Choose the correct answer from the options given below.

- 1) (a) - (ii), (b) - (iv), (c) - (i), (d) - (iii)
 2) (a) - (ii), (b) - (iv), (c) - (iii), (d) - (i)
 3) (a) - (iv), (b) - (ii), (c) - (i), (d) - (iii)
 4) (a) - (ii), (b) - (i), (c) - (iv), (d) - (iii)

Ans. 1

Sol. Conceptual

40. Given below are two statements: One is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A):

The stretching of a spring is determined by the shear modulus of the material of the spring.

Reason (R):

A coil spring of copper has more tensile strength than a steel spring of same dimensions.

In the light of the above statements, choose the most appropriate answer from the options given below.

- 1) Both (A) and (R) are true and (R) is not the correct explanation of (A)
 2) (A) is true but (R) is false
 3) (A) is false but (R) is true
 4) Both (A) and (R) are true and (R) is the correct explanation (A)

Ans. 2

Sol. Conceptual

41. The area of rectangular field (in m^2) of length 55.3m and breadth 25m after rounding off the value for correct significant digits is:

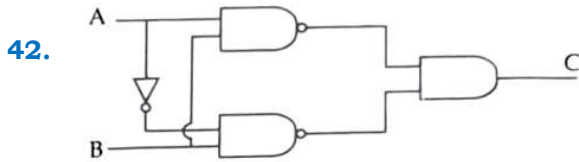
- 1) 1382 2) 1382.5
 3) 14×10^2 4) 138×10^1

Ans. 3

Sol.
$$A = \ell \times b$$

$$= 55.3 \times 25 = 1382.5$$
 On rounding off for two significant figures

$$= 14 \times 10^2 \text{ m}$$



The truth table for the given logic circuit is:

A	B	C
0	0	1
0	1	0
1	0	0
1	1	1

A	B	C
0	0	1
0	1	0
1	0	1
1	1	0

A	B	C
0	0	0
0	1	1
1	0	0
1	1	1

A	B	C
0	0	0
0	1	1
1	0	1
1	1	0

Ans. 2

Sol. Conceptual

43. The volume occupied by the molecules contained in 4.5kg water at STP, if the intermolecular forces vanish away is:

- 1) $5.6 \times 10^3 \text{ m}^3$ 2) $5.6 \times 10^{-3} \text{ m}^3$
 3) 5.6 m^3 4) $5.6 \times 10^6 \text{ m}^3$

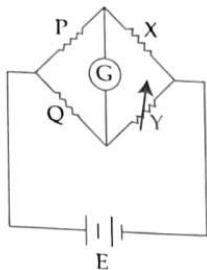
Ans. 3

Sol. $m = 4.5 \text{ kg}$

$$PV = nRT$$

$$V = \frac{nRT}{P} = \frac{250 \times 8.314 \times 273}{10^5} = 5.67 \text{ m}^3$$

44. A Wheatstone bridge is used to determine the value of unknown resistance X by adjusting the variable resistance Y as shown in the figure. For the most precise measurement of X, the resistances P and Q:



- 1) should be approximately equal and are small
 2) should be very large and unequal
 3) do not play any significant role
 4) should be approximately equal to 2X

Ans. 1

Sol. Conceptual

45. A series LCR circuit with inductance 10 H, capacitance $10 \mu\text{F}$, resistance 50Ω is connected to an ac source of voltage, $V = 200 \sin(100t)$ volt. If the resonant frequency of the LCR circuit is ν_0 and the frequency of the ac source is ν , then:

- 1) $\nu_0 = \nu = \frac{50}{\pi} \text{ Hz}$
 2) $\nu_0 = \frac{50}{\pi} \text{ Hz}$, $\nu_0 = 50 \text{ Hz}$
 3) $\nu = 100 \text{ Hz}$, $\nu_0 = \frac{100}{\pi} \text{ Hz}$
 4) $\nu_0 = \nu = 50 \text{ Hz}$

Ans. 1

Sol. $L = 10 \text{ H}$

$$C = 10 \mu\text{F}$$

$$R = 50 \Omega$$

$$V = 200 \sin(100t)$$

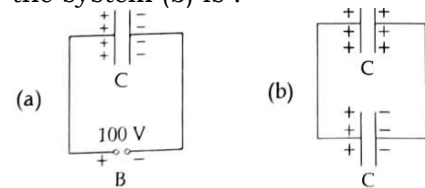
$$\nu_0 = \frac{1}{2\pi \sqrt{LC}}$$

$$\nu_0 = \frac{1}{2\pi \sqrt{10 \times 10 \times 10^{-6}}} = \frac{10^2}{2\pi} = \frac{50}{\pi}$$

$$100 = 2\pi\nu$$

$$\nu = \frac{100}{2\pi} = \frac{50}{\pi}$$

46. A capacitor of capacitance $C = 900 \text{ pF}$ is charged fully by 100 V battery B as shown in figure (a). Then it is disconnected from the battery and connected to another uncharged capacitor of capacitance $C = 900 \text{ pF}$ as shown in figure (b). The electrostatic energy stored by the system (b) is :



- 1) $3.25 \times 10^{-6} \text{ J}$ 2) $2.25 \times 10^{-6} \text{ J}$
 3) $1.5 \times 10^{-6} \text{ J}$ 4) $4.5 \times 10^{-6} \text{ J}$

Ans. 2

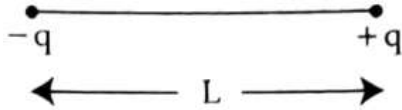
Sol. $V = \frac{C_1 V_1 + C_2 V_2}{C_1 + C_2} = \frac{900 \times 100}{2 \times 900} = 50$

$$V = \frac{1}{2} (C_1 + C_2) V^2$$

$$V = \frac{1}{2} \times 1800 \times 10^{-12} \times (50^2)$$

$$V = 900 \times 10^{-12} \times 2500 = 2.25 \times 10^{-6} \text{ J}$$

47. Two point charges $-q$ and $+q$ are placed at a distance of L , as shown in the figure.



The magnitude of electric field intensity at a distance R ($R \gg L$) varies as:

- 1) $\frac{1}{R^3}$ 2) $\frac{1}{R^4}$ 3) $\frac{1}{R^6}$ 4) $\frac{1}{R^2}$

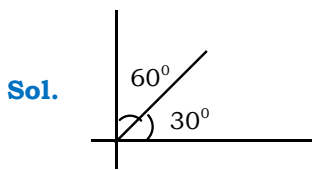
Ans. 1

Sol. $\frac{-q}{\quad} \quad \frac{+q}{\quad}$
 $E \propto \frac{1}{R^3}$

48. A ball is projected with a velocity, 10 ms^{-1} , at an angle of 60° with the vertical direction. Its speed at the highest point of its trajectory will be:

- 1) $5\sqrt{3} \text{ ms}^{-1}$ 2) 5 ms^{-1}
 3) 10 ms^{-1} 4) Zero

Ans. 1



Sol.

$u_x = u \cos \theta$
 $u_x = 10 \cos 30 = 10 \frac{\sqrt{3}}{2} = 5\sqrt{3}$

49. From Ampere's circuital law for a long straight wire of circular cross-section carrying a steady current, the variation of magnetic field in the inside and outside region of the wire is:

- 1) a linearly increasing function of distance upto the boundary of the wire and then linearly decreasing for the outside region.
 2) a linearly increasing function of distance r upto the boundary of the wire and then decreasing on with $1/r$ dependence for the outside region.
 3) a linearly decreasing function of distance upto the boundary of the wire and then a linearly increasing one for the outside region
 4) uniform and remains constant for both the regions.

Ans. 2

Sol. Conceptual

50. Two pendulums of length 121 cm and 100 cm start vibrating in phase. At some instant, the two are at their mean position in the same phase. The minimum number of vibrations of the shorter pendulum after which the two are again in phase at the mean position is:

- 1) 9 2) 10 3) 8 4) 11

Ans. 4

Sol. $(n-1)\sqrt{l_1} = n\sqrt{l_2}$
 $(n-1)11 = n10$
 $11n - 11 = 10n$
 $n = 11$



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NEET EXAMINATION – 2022

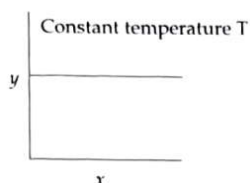
VERSION :
T6

DATE :- 17-07-2021

TIME : 02.00 PM TO 05.20 PM

CHEMISTRY - SECTION - A

51. The given graph is a representation of kinetics of a reaction.



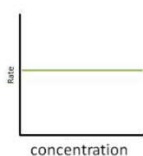
The y and x axes for zero and first order reactions, respectively are

- 1) zero order (y=concentration and x=time) first order (y=rate constant and x=concentration)
- 2) zero order (y=rate and x=concentration), first order (y= $t_{1/2}$ and x= concentration)
- 3) zero order (y=rate and x= concentration) first order (y=rate and x= $t_{1/2}$)
- 4) zero order (y=concentration and x=time), first order (y= $t_{1/2}$ and x=concentration)

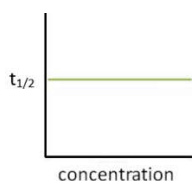
Ans. 2

Sol. For zero order reaction

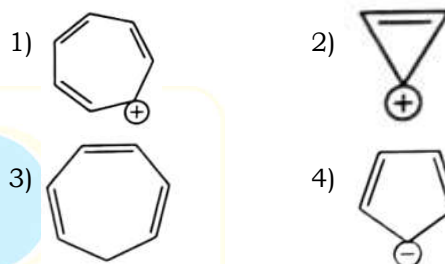
$$\text{Rate} \propto a^0$$



For 1st order reaction $t_{1/2} \propto a^0$

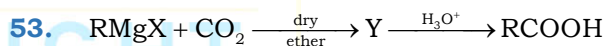


52. Which compound amongst the following is not an aromatic compound?



Ans. 3

Sol. 1, 2 and 4 compounds are aromatic and 3 compound is non-aromatic



What is Y in the above reaction?

- 1) $\text{R}_3\text{CO}^-\text{Mg}^+\text{X}$
- 2) RCOO^-X^+
- 3) $(\text{RCOO})_2\text{Mg}$
- 4) $\text{RCOO}^-\text{Mg}^+\text{X}$

Ans. 4

Sol. $\text{Y} = \text{R} - \text{COO}^-\text{Mg}^+\text{x}$

54. Which of the following statement is not correct about diborane?

- 1) The four terminal B-H bonds are two centre two electron bonds.
- 2) The four terminal Hydrogen atoms and the two Boron atoms lie in one plane.
- 3) Both the Boron atoms are sp^2 hybridised.
- 4) There are two 3-centre-2-electron bonds.

Ans. 3

Sol. in diborane both borons undergoes sp^3 hybridization

55. Gadolinium has low value of third ionisation enthalpy because of
- 1) high exchange enthalpy
 - 2) high electronegativity
 - 3) high basic character
 - 4) small size

Ans. 1

Sol. ${}_{64}\text{Gd}$ Electronic configuration is $4s^7 5d^1 6s^2$
Half filled Electronic configuration is will exhibit more exchange enthalpy

56. Which one is not correct mathematical equation for Dalton's Law of partial pressure? Here p = total pressure of gaseous mixture.

1) $p = n_1 \frac{RT}{V} + n_2 \frac{RT}{V} + n_3 \frac{RT}{V}$

2) $p_i = x_i p$, where

p_i = partial pressure of i^{th} gas

x_i = mole fraction of i^{th} gas in gaseous mixture

3) $p_i = x_i p_i^0$, where

x_i = mole fraction of i^{th} gas in gaseous mixture

p_i^0 = pressure of i^{th} gas in pure state

4) $p = p_1 + p_2 + p_3$

Ans. 3

Sol. Partial pressure of i^{th} gas in gaseous mixture
 $p_i = X_i P$

57. Given below are two statements:

Statement-I:

The boiling points of the following hydrides of group 16 elements increases in the order:
 $\text{H}_2\text{O} < \text{H}_2\text{S} < \text{H}_2\text{Se} < \text{H}_2\text{Te}$.

Statement-II:

The boiling points of these hydrides increase with increase in molar mass.

In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement-I and Statement-II are incorrect
- 2) Statement-I is correct but Statement-II is incorrect
- 3) Statement-I is incorrect but Statement-II is correct
- 4) Both Statement-I and Statement-II are correct

Ans. 1

Sol. BP order $\text{H}_2\text{S} < \text{H}_2\text{Se} < \text{H}_2\text{Te} < \text{H}_2\text{O}$
 213K 232K 269K 373K

due to H-bond H_2O boiling point is more

58. Given below are two statements:

Statement-I:

The acidic strength of monosubstituted nitrophenol is higher than phenol because of electron withdrawing nitro group.

Statement-II:

o-nitrophenol, m-nitrophenol and p-nitrophenol will have same acidic strength as they have one nitro group attached to the phenolic ring.

In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement-I and Statement-II are incorrect
- 2) Statement-I is correct but Statement-II is incorrect
- 3) Statement-I is incorrect but Statement-II is correct
- 4) Both Statement-I and Statement-II are correct

Ans. 2

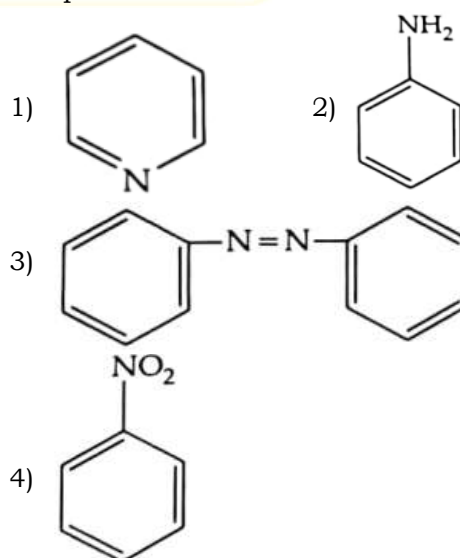
Sol. Statement I: Acidic strength order:

Nitrophenol > Phenol

Statement II: Acidic strength order

Paranitrophenol > orthonitrophenol > metanitrophenol

59. The Kjeldhal's method for the estimation of nitrogen can be used to estimate the amount of nitrogen in which one of the following compounds?



Ans. 2

Sol. Kjeldahl's method is not applicable to compounds containing nitrogen in **Nitro** and **Azo** groups and **nitrogen in the ring**

60. Match List-I with List-II.

List-I (Drug class)	List-II (Drug molecule)
(a) Antacids	(i) Salvarsan
(b) Antihistamines	(ii) Morphine
(c) Analgesics	(iii) Cimetidine
(d) Antimicrobials	(iv) Seldane

Choose the correct answer from the options given below:

- 1) (a) - (iii), (b) - (iv), (c) - (ii), (d) - (i)
- 2) (a) - (i), (b) - (iv), (c) - (ii), (d) - (iii)
- 3) (a) - (iv), (b) - (iii), (c) - (i), (d) - (ii)
- 4) (a) - (iii), (b) - (ii), (c) - (iv), (d) - (i)

Ans. 1

Sol. Antacid - cimetidine

Antihistamine - seldane

Analgesics - Morphine

Antimicrobials - salvarsan

61. The IUPAC name of an element with atomic number 119 is

- 1) unnilennium
- 2) unununnium
- 3) ununoctium
- 4) ununennium

Ans. 4

Sol. Z= 119 ununennium

62. Amongst the following which one will have maximum 'lone pair-lone pair' electron repulsions?

- 1) IF_5
- 2) SF_4
- 3) XeF_2
- 4) ClF_3

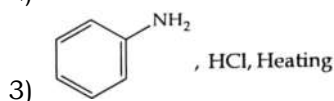
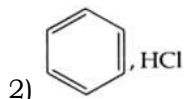
Ans. 3

Sol. According to VSEPR theory $L.P-L.P > B.P-L.P > B.P-B.P$

In XeF_2 three lonepairs on the central atom and causes more repulsions

63. Which of the following sequence of reactions is suitable to synthesize chlorobenzene?

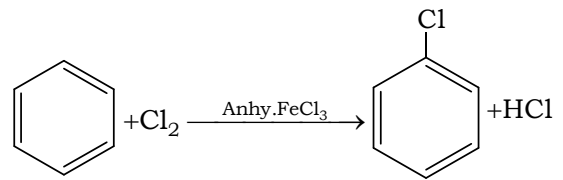
- 1) Phenol, NaNO_2 , HCl , CuCl



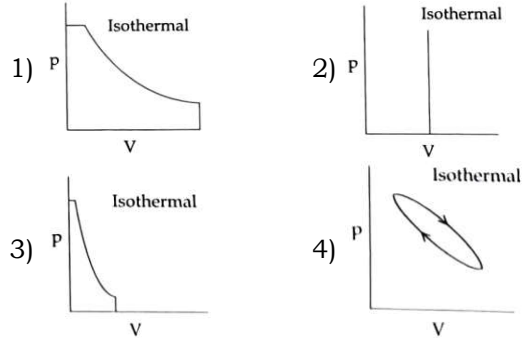
- 4) Benzene, Cl_2 , anhydrous FeCl_3

Ans. 4

Sol. Halogenation

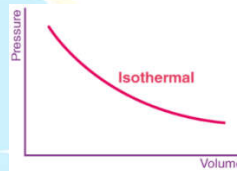


64. Which of the following p-V curve represents maximum work done?

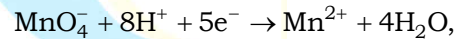


Ans. 1

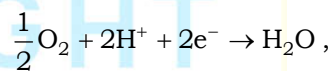
Sol. Maximum work done in Reversible isothermal process



65. Given below are half cell reactions:



$$E_{\text{Mn}^{2+}/\text{MnO}_4^-}^\circ = -1.510\text{V}$$



$$E_{\text{O}_2/\text{H}_2\text{O}}^\circ = +1.223\text{V}$$

Will the permanganate ion, MnO_4^- liberate O_2 from water in the presence of an acid?

- 1) No, because $E_{\text{cell}}^\circ = -0.287\text{V}$
- 2) Yes, because $E_{\text{cell}}^\circ = +2.733\text{V}$
- 3) No, because $E_{\text{cell}}^\circ = -2.733\text{V}$
- 4) Yes, because $E_{\text{cell}}^\circ = +0.287\text{V}$

Ans. 4

Sol. Given SOP $E_{\text{Mn}^{2+}/\text{MnO}_4^-}^\circ = -1.510$
 \Rightarrow SRP $E_{\text{MnO}_4^-/\text{Mn}^{2+}}^\circ = +1.510$ acts as cathode

and SRP $E_{\text{O}_2/\text{H}_2\text{O}}^\circ = +1.223\text{V}$ acts as anode

$$E_{\text{cell}}^\circ = E_{\text{RHS}}^\circ - E_{\text{LHS}}^\circ = 1.510 - 1.223 = +0.287\text{V}$$

66. The IUPAC name of the complex $[\text{Ag}(\text{H}_2\text{O})_2][\text{Ag}(\text{CN})_2]$ is:

- 1) diaquasilver (II) dicyanidoargentate (II)
- 2) dicyanidosilver (I) diaquaargentate(I)
- 3) diaquasilver (I) dicyanidoargentate(I)
- 4) dicyanidosilver (II) diaquaargentate(II)

Ans. 3

Sol. Diaquasilver (I) dicyanidoargentate(I)

67. Identify the incorrect statement from the following

- 1) The oxidation number of K in KO_2 is +4
- 2) Ionisation enthalpy of alkali metals decreases from top to bottom in the group.
- 3) Lithium is the strongest reducing agent among the alkali metals
- 4) Alkali metals react with water to form their hydroxides.

Ans. 1

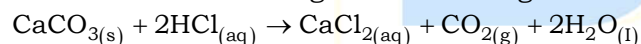
Sol. KO_2 potassium superoxide ($\text{K}^+\text{O}_2^{-1}$)

Potassium oxidation number = +1

Or

IA group elements oxidation number is = +1

68. What mass of 95% pure CaCO_3 will be required to neutralise 50mL of 0.5M HCl solution according to the following reaction ?



[Calculate upto second place of decimal point]

- 1) 1.32g
- 2) 3.65g
- 3) 9.50g
- 4) 1.25g

Ans. 1

Sol. Meq of CaCO_3 = meq of HCl

$$\frac{\text{wt}}{\text{Eq.wt}} \times 1000 = M \times n_f \times V_{\text{ml}}$$

$$\text{wt} = \frac{M \times n_f \times V_{\text{ml}} \times \text{Eq.wt}}{1000}$$

$$\text{wt} = \frac{0.5 \times 1 \times 50 \times 50}{1000}$$

$$\text{wt} = 1.25\text{g}$$

$$95\% \text{ of pure } \text{CaCO}_3 \rightarrow \frac{100 \times 1.25}{95} = 1.32\text{g}$$

69. Given below are two statements:

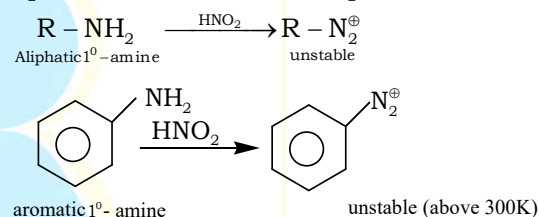
Statement I: Primary aliphatic amines react with HNO_2 to give unstable diazonium salts.

Statement II: Primary aromatic amines react with HNO_2 to form diazonium salts which are stable even above 300K. In the light of the above statements, choose the most appropriate answer from the option given below:

- 1) Both statement I and Statement II are incorrect
- 2) Statement I is correct but statement II is incorrect.
- 3) Statement I is incorrect but Statement II is correct.
- 4) Both Statement I and Statement II are correct

Ans. 2

Sol. Statement I is correct, Statement II is incorrect because aromatic primary amines are stable up to zero to 5°C that is up to 278°K



70. Which statement regarding polymers is not correct?

- 1) Fibers possess high tensile strength
- 2) Thermoplastic polymers are capable of repeatedly softening and hardening on heating cooling respectively.
- 3) Thermosetting polymers are reusable.
- 4) Elastomers have polymer chains held together by weak intermolecular forces

Ans. 3

Sol. Thermosetting polymers cannot be reused

71. The incorrect statement regarding chirality is:

- 1) The product obtained by $\text{S}_{\text{N}}2$ reaction of haloalkane having chirality at the reactive site shows inversion of configuration.
- 2) Enantiomers are superimposable mirror images on each other.
- 3) A racemic mixture shows zero optical rotation
- 4) $\text{S}_{\text{N}}1$ reaction yields 1:1 mixture of both enantiomers

Ans. 2

Sol. Enantiomers are Non-superimposable mirror images on each other.

72. Given below are two statements:

Statement I:

In the coagulation of a negative sol, the flocculating power of the three given ions is in the order-



Statement II:

In the coagulation of a positive sol, the flocculating power of the three given salts is in the order-



In the light of the above statements, choose the most appropriate answer form the options given below:

- 1) Both Statements I and Statements II are incorrect
- 2) Statements I is correct but Statements II is incorrect
- 3) Statements I is incorrect but Statements II is correct.
- 4) Both Statements I and Statements II are correct.

Ans. 2

Sol. In the coagulation of a -ve sol, the flocculating power $\text{Al}^{+3} > \text{Ba}^{+2} > \text{Na}^{+1}$
 In the coagulation of a +ve sol, the flocculating power $\text{PO}_4^{3-} > \text{SO}_4^{2-} > \text{Cl}^-$

73. Match List-I with List-II.

	List-I (Hydrides)		List-II (Nature)
a)	MgH ₂	i)	Electron precise
b)	GeH ₄	ii)	Electron deficient
c)	B ₂ H ₆	iii)	Electron rich
d)	HF	iv)	Ionic

Choose the correct answer form the options given below:

- 1) (a) – (iii), (b) – (i), (c) – (ii), (d) – (iv)
- 2) (a) – (i), (b) – (ii), (c) – (iv), (d) – (iii)
- 3) (a) – (ii), (b) – (iii), (c) – (iv), (d) – (i)
- 4) (a) – (iv), (b) – (i), (c) – (ii), (d) – (iii)

Ans. 4

Sol. MgH₂ : IONIC, but predominantly covalent in Nature
 GeH₄ : Electron precise

B₂H₆ : Electron deficient

HF : Electron rich

74. Match List-I with List-II

	List-I		List-II
a)	Li	i)	Absorbent for carbon dioxide
b)	Na	ii)	Electrochemical cells
c)	KOH	iii)	Coolant in fast breeder reactors
d)	Cs	iv)	Photoelectric cell

Choose the correct answer form the options given below:

- 1) (a) – (iii), (b) – (iv), (c) – (ii), (d) – (i)
- 2) (a) – (i), (b) – (iii), (c) – (iv), (d) – (ii)
- 3) (a) – (ii), (b) – (iii), (c) – (i), (d) – (iv)
- 4) (a) – (iv), (b) – (i), (c) – (iii), (d) – (ii)

Ans. 3

Sol. Li : Electrochemical cells
 Na : Coolant in fast breeder reactions
 KOH : Absorbent for CO₂
 Cs : Photoelectric cell

75. Given below are two statements: one is labelled as Assertion (A) and the other is labelled as Reason (R)

Assertion (A)

In a particular point defect, an ionic solid is electrically neutral, even if few of its cations are missing from its unit cells.

Reason (R)

In a ionic solid, Frenkel defect arises due to dislocation of cation form its lattice site to interstitial site, maintaining overall electrical neutrality.

In the light of the above statements, choose the most appropriate answer form the options given below:

- 1) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- 2) (A) is correct but (R) is not correct
- 3) (A) is not correct but (R) is correct
- 4) Both (A) and (R) are correct and (R) is the correct explanation of (A)

Ans. 4

Sol. Conceptual

76. The incorrect statement regarding enzymes is:
- 1) Like chemical catalysts enzymes reduce the activation energy of bio processes
 - 2) Enzymes are polysaccharides
 - 3) Enzymes are very specific for a particular reaction and substrate
 - 4) Enzymes are biocatalysts

Ans. 2

Sol. Conceptual

77. Match List-I with List-II

	List-I (Products formed)		List-II (Reaction of carbonyl compound with)
a)	Cyanohydrin	i)	NH ₂ OH
b)	Acetal	ii)	RNH ₂
c)	Schiff's base	iii)	Alcohol
d)	Oxime	iv)	HCN

Choose the correct answer from the options given below:

- 1) (a) – (ii), (b) – (iii), (c) – (iv), (d) – (i)
- 2) (a) – (i), (b) – (iii), (c) – (ii), (d) – (iv)
- 3) (a) – (iv), (b) – (iii), (c) – (ii), (d) – (i)
- 4) (a) – (iii), (b) – (iv), (c) – (ii), (d) – (i)

Ans. 3

Sol. Conceptual

78. In one molal solution that contains 0.5 mole of a solute, there is
- 1) 500g of solvent
 - 2) 100 mL of solvent
 - 3) 1000g of solvent
 - 4) 500 mL of solvent

Ans. 1

Sol. $m = n \times \frac{1000}{\text{wt of solvent}}$

$$\text{wt of solvent} = n \times \frac{1000}{m}$$

$$0.5 \times \frac{1000}{1} = 500$$

$$\text{wt} = 500\text{g of solvent}$$

79. Choose the correct statement:

- 1) Diamond is covalent and graphite is ionic.
- 2) Diamond is sp³ hybridised and graphite is sp² hybridized.
- 3) Both diamond and graphite are used as dry lubricants
- 4) Diamond and graphite have two dimensional network

Ans. 2

Sol. Conceptual

80. Which amongst the following is incorrect statement?

- 1) C₂ molecule has four electrons in its two degenerate π molecular orbitals.
- 2) H₂⁺ ions has one electron
- 3) O₂⁺ ion is diamagnetic
- 4) The bond orders of O₂⁺, O₂, O₂⁻ and O₂²⁻ are 2.5, 2, 1.5 and 1, respectively

Ans. 3

Sol. Conceptual

81. Given below are two statements: One is labelled as

Assertion (A) and the other is labelled as Reason (R)

Assertion (A): ICl is more reactive than I₂

Reason(R): I-Cl bond is weaker than I-I bond.

In the light of the above statements, choose the most appropriate answer from the options below

- 1) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- 2) (A) is correct but (R) is not correct
- 3) (A) is not correct but (R) is correct
- 4) Both (A) and (R) are correct and (R) is the correct explanation of (A)

Ans. 4

Sol. Conceptual

82. Given below are two statements:

Statement I: The boiling points of aldehydes and ketones are higher than hydrocarbons of comparable molecular masses because of weak molecular association in aldehydes and ketones due to dipole-dipole interactions

Statement II: The boiling points of aldehydes and ketones are lower than the alcohols of similar molecular masses due to the absence of H-bonding

In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect.
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both statement I and Statement II are correct

Ans. 3

Sol. Conceptual

83. The pH of the solution containing 50mL each of 0.10M sodium acetate and 0.01 M acetic acid is

[Given pK_a of $CH_3COOH = 4.57$]

- 1) 3.57
- 2) 4.57
- 3) 2.57
- 4) 5.57

Ans. 4

Sol.
$$p^H = p^{ka} + \log \left[\frac{\text{salt}}{\text{Acid}} \right]$$

$$p^H = 4.57 + \log \left[\frac{0.10}{0.01} \right]$$

$$= 4.57 + 1$$

$$p^H = 5.57$$

84. Identify the incorrect statement from the following

- 1) All the five 4d orbitals have shapes similar to the respective 3d orbitals.
- 2) In an atom, all the five 3d orbitals are equal in energy in free state.
- 3) The shapes of d_{xy} , d_{yz} , and d_{zx} orbitals are similar to each other and $d_{x^2-y^2}$ and d_{z^2} are similar to each other
- 4) All the five 5d orbitals are different in size when compared to the respective 4d orbitals

Ans. 3

Sol. Shapes of d_{xy} , d_{yz} , d_{zx} are similar to each other but $d_{x^2-y^2}$ but d_{z^2} is not similar each other.

85. At 298K, the standard electrode potentials of Cu^{2+} / Cu , Zn^{2+} / Zn , Fe^{2+} / Fe and Ag^+ / Ag are 0.34V, -0.76V, -0.44V and 0.80V, respectively. On the basis of standard electrode potential, predict Which of the following reaction can not occur?

- 1) $CuSO_4(aq) + Fe(s) \rightarrow FeSO_4(aq) + Cu(s)$
- 2) $FeSO_4(aq) + Zn(s) \rightarrow ZnSO_4(aq) + Fe(s)$
- 3) $2CuSO_4(aq) + 2Ag(s) \rightarrow 2Cu(s) + Ag_2SO_4(aq)$
- 4) $CuSO_4(aq) + Zn(s) \rightarrow ZnSO_4(aq) + Cu(s)$

Ans. 3

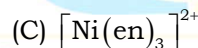
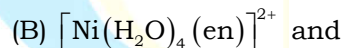
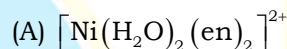
Sol. $Cu^{+2} / Cu = 0.34v$ R.A

$$Ag^+ / Ag = 0.80v \quad O.A$$

Cu has low S.R.P so it reduces $Ag^+ \rightarrow Ag$

CHEMISTRY - SECTION - B

86. The order of energy absorbed which is responsible for the colour of complexes



1) (C) > (B) > (A) 2) (C) > (A) > (B)

3) (B) > (A) > (C) 4) (A) > (B) > (C)

Ans. 2

Sol. Energy absorbed \propto strength of ligand
 $C > A > B$

87. A 10.0L flask contains 64 g of oxygen at 27°C. (Assume O_2 gas is behaving ideally). The pressure inside the flask in bar is (Given $R=0.0831 \text{ L bar K}^{-1} \text{ mol}^{-1}$)

- 1) 498.6
- 2) 49.8
- 3) 4.9
- 4) 2.5

Ans. 3

Sol. $V = 10 \text{ Lit}$

$$W_{O_2} = 64 \text{ gm} \quad M. \text{wt} = 32$$

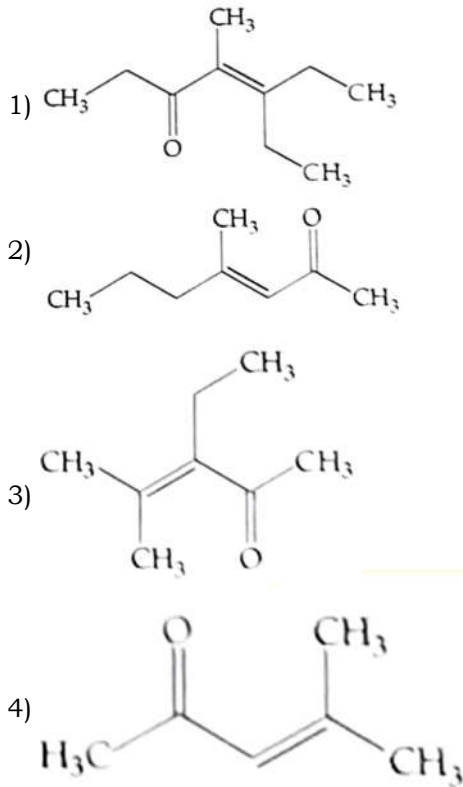
$$T = 27^\circ C = 300K$$

$$P = ?$$

$$PV = \frac{w RT}{M}$$

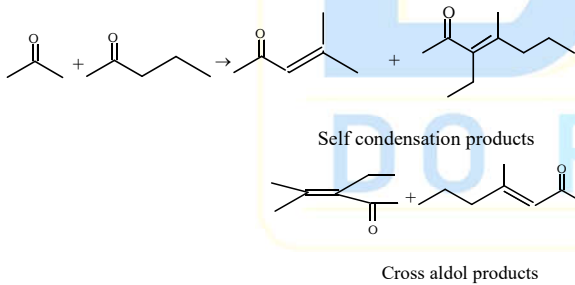
$$PV = \frac{64 \times 0.0831 \times 300}{10 \times 32} = 4.9$$

88. Which one of the following is not formed when acetone reacts with 2-pentanone in the presence of dilute NaOH followed by heating

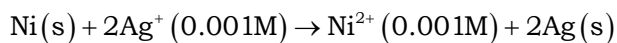


Ans. 1

Sol. Cross-aldol condensation



89. Find the emf of the cell in which the following reaction takes place at 298K

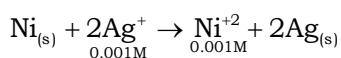


(Given that $E_{\text{cell}}^0 = 10.5\text{V}$, $\frac{2.303RT}{F} = 0.059$ at 298K)

1) 1.385V 2) 0.9615V 3) 1.05V 4) 1.0385V

Ans. NO OPTION

Sol. $E_{\text{cell}}^0 = 10.5$



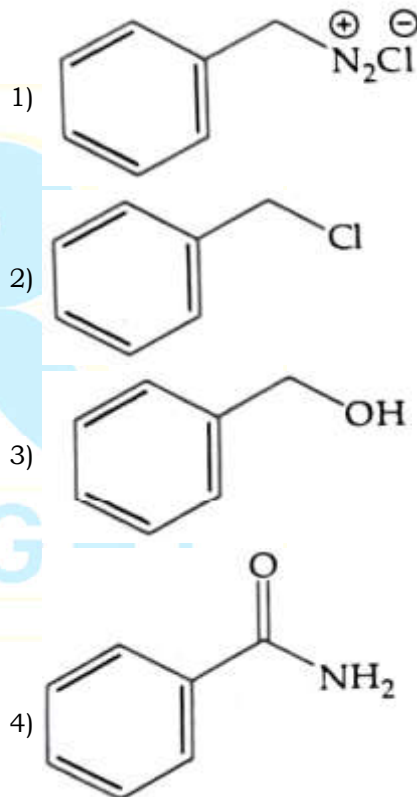
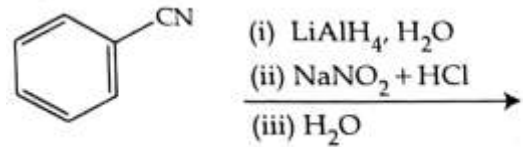
$$E_{\text{cell}} = E_{\text{cell}}^0 - \frac{0.059}{2} \log \frac{[\text{Ni}^{2+}]}{[\text{Ag}^+]^2}$$

$$E_{\text{cell}} = 10.5 - \frac{0.059}{2} \log \frac{0.001}{(0.001)^2}$$

$$E_{\text{cell}} = 10.5 - \frac{0.059 \times 3}{2} = 9.615$$

(Grace: E_{cell}^0 given as 10.5V **but correct is 1.05V so correct answer is 0.9615**)

90. The product formed from the following reaction sequence is



Ans. 3

Sol. Conceptual

91. Copper crystallises in fcc unit cell with cell edge length of 3.608×10^{-8} cm. The density of copper of 8.92g cm^{-3} . Calculate the atomic mass of copper.

1) 31.55 u 2) 60 u
3) 65 u 4) 63.1 u

Ans. 4

Sol. FCC Z=4
a = 3.608×10^{-8} cm
d = 8.92g cm^{-3}

At weight of Cu (M)=?

$$d = \frac{ZM}{N_A a^3}$$

$$M = \frac{8.92 \times 6.023 \times 10^{23} \times (3.608 \times 10^{-8})^3}{4}$$

$$M = 63.1$$

92. $3O_2(g) \rightleftharpoons 2O_3(g)$ for the above reaction at 298 K, K_C is found to be 3.0×10^{-59} . If the concentration of O_2 at equilibrium is 0.040 M then concentration of O_3 in M is

- 1) 1.9×10^{-63} 2) 2.4×10^{31}
3) 1.2×10^{21} 4) 4.38×10^{-32}

Ans. 4

Sol. $3O_{2(g)} \rightleftharpoons 2O_{3(g)}$ $K_c = 3 \times 10^{-59}$

$$[O_2] = 0.04 \text{ M}$$

$$[O_3] = ?$$

$$K_c = \frac{[O_3]^2}{[O_2]^3}$$

$$3 \times 10^{-59} = \frac{[O_3]^2}{[0.04]^3}$$

$$[O_3]^2 = 3 \times 10^{-59} \times 64 \times 10^{-6}$$

$$[O_3]^2 = 19.2 \times 10^{-64}$$

$$[O_3]^2 = \sqrt{19.2 \times 10^{-64}} = 4.38 \times 10^{-32}$$

93. Given below are two statements:
Statement I : In Lucas test, primary, secondary and tertiary alcohols are distinguished on the basis of their reactivity with conc. $HCl + ZnCl_2$, known as Lucas Reagent.

Statement II: Primary alcohols are most reactive and immediately produce turbidity at room temperature on reaction with Lucas Reagent.

In the light of the above statement, choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
2) Statement I is correct but statement II is incorrect
3) Statement I is incorrect but statement II is correct
4) Both statement I and Statement II are correct

Ans. 2

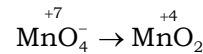
Sol. Conceptual

94. In the neutral or faintly alkaline medium, $KMnO_4$ oxidises iodide into iodate. The change in oxidation state of manganese in this reaction is from

- 1) +6 to +4 2) +7 to +3
3) +6 to +5 4) +7 to +4

Ans. 4

Sol. $2MnO_4^- + H_2O + I^- \rightarrow 2MnO_2 + 2OH^- + IO_3^-$



95. Math List – I with List – II.

	List – I (Ores)		List – II (Composition)
(a)	Haematite	(i)	Fe_3O_4
(b)	Magnetite	(ii)	$ZnCO_3$
(c)	Calamine	(iii)	Fe_2O_3
(d)	Kaolinite	(iv)	$[Al_2(OH)_4Si_2O_5]$

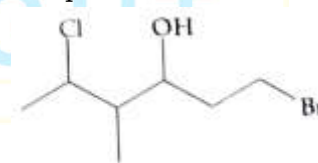
Choose the correct answer from the options given below:

- 1) (a) – (iii), (b) – (i), (c) – (ii), (d) – (iv)
2) (a) – (iii), (b) – (i), (c) – (iv), (d) – (ii)
3) (a) – (i), (b) – (iii), (c) – (ii), (d) – (iv)
4) (a) – (i), (b) – (ii), (c) – (iii), (d) – (iv)

Ans. 1

Sol. Conceptual

96. The correct IUPAC name of the following compound is:



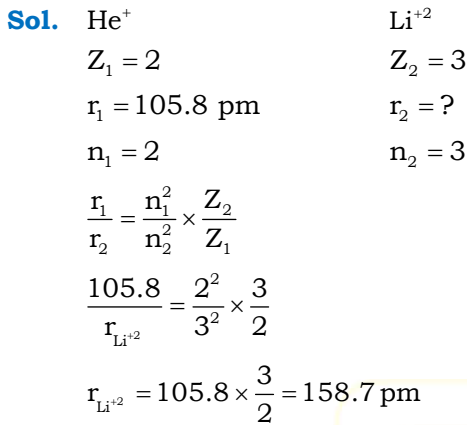
- 1) 6-bromo-2-chloro-4-methylhexan-4-ol
2) 1-bromo-4-methyl-5-chlorohexan-3-ol
3) 6-bromo-4-methyl-2-chlorohexan-4-ol
4) 1-bromo-5-chloro-4-methylhexan-3-ol

Ans. 4

Sol. Conceptual

97. If radius of second Bohr orbit of the He^+ ion is 105.8 pm, what is the radius of third Bohr orbit of Li^{2+} ions?
- 1) 15.87 pm 2) 1.587 pm
 3) $158.7 \overset{\circ}{\text{A}}$ 4) 158.7 pm

Ans. 4



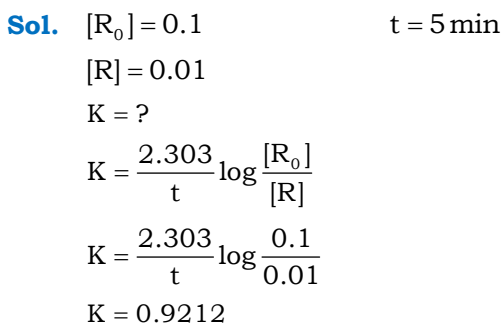
98. Compound X on reaction with O_3 followed by $\text{Zn}/\text{H}_2\text{O}$ gives formaldehyde and 2-methyl propanal as products. The compound X is:
- 1) 2-Methylbut-1-ene 2) 2-Methylbut-2-ene
 3) Pent-2-ene 4) 3-Methylbut-1-ene

Ans. 4

Sol. Conceptual

99. For a first order reaction $\text{A} \rightarrow \text{products}$, initial concentration of A is 0.1 M, which becomes 0.001 M after 5 minutes. Rate constant for the reaction in min^{-1} is
- 1) 0.9212 2) 0.4606
 3) 0.2303 4) 1.3818

Ans. 1



100. The pollution due to oxides of sulphur gets enhanced due to the presence of:
- a) Particulate matter b) Ozone
 c) hydrocarbons d) hydrogen peroxide
- Choose the most appropriate answer from the options given below:

- 1) (a), (b), (d) only 2) (b), (c), (d) only
 3) (a), (c), (d) only 4) (a), (d) only

Ans. 1

Sol. Conceptual



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NEET EXAMINATION – 2022

VERSION :
T6

DATE :- 17-07-2021

TIME : 02.00 PM TO 05.20 PM

BOTANY - SECTION - A

101. Which of the following is not a method of *ex situ* conservation?

- 1) National Parks 2) Micropropagation
3) Cryopreservation 4) *In vitro* fertilization

Ans. 1

102. Given below are two statements:

Statement I: The primary CO₂ acceptor in C₄ plants is phosphoenolpyruvate and is found in the mesophyll cells.

Statement II: Mesophyll cells of C₄ plants lack RuBisCo enzyme. In the light of the above statements, choose the correct answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
2) Statement I is correct but statement II is incorrect
3) Statement I is incorrect but statement II is correct
4) Both statement I and Statement II are correct

Ans. 4

103. XO type of sex determination can be found in:

- 1) Birds 2) Grasshoppers
3) Monkeys 4) *Drosophila*

Ans. 2

104. In old trees the greater part of secondary xylem is dark brown and resistant to insect attack due to:

- a) secretion of secondary metabolites and their deposition in the lumen of vessels.
b) deposition of organic compounds like tannins and resins in the central layers of stem.
c) deposition of suberin and aromatic substances in the outer layer of stem.
d) deposition of tannins, gum, resin and aromatic substances in the peripheral layers of stem.
e) presence of parenchyma cells, functionally active xylem elements and essential oils.

Choose the correct answer from the options given below:

- 1) (c) and (d) only 2) (d) and (e) only
3) (b) and (d) only 4) (a) and (b) only

Ans. 4

105. Which of the following is not observed during apoplastic pathway?

- 1) The movement does not involve crossing of cell membrane
2) The movement is aided by cytoplasmic streaming
3) Apoplast is continuous and does not provide any barrier to water movement.
4) Movement of water occurs through intercellular spaces and wall of the cells.

Ans. 2

106. Given below are two statements: One is labelled as **Assertion (A)** and the other is labelled as **Reason (R)**.

Assertion (A): Polymerase chain reaction is used in DNA amplification

Reason (R): The ampicillin resistant gene is used as a selectable marker to check transformation

In the light of the above statements, choose the correct answer from the options given below:

- 1) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- 2) (A) is correct but (R) is not correct
- 3) (A) is not correct but (R) is correct
- 4) Both (A) and (R) are correct and (R) is the correct explanation of (A)

Ans. 1

107. Which one of the following statement is not true regarding gel electrophoresis technique?

- 1) The separated DNA fragments are stained by using ethidium bromide.
- 2) The presence of chromogenic substrate gives blue coloured DNA bands on the gel.
- 3) Bright orange coloured bands of DNA can be observed in the gel when exposed to UV light.
- 4) The process of extraction separated DNA strands from gel is called elution.

Ans. 2

108. The gaseous plant growth regulator is used in plants to:

- 1) promote root growth and root hair formation to increase the absorption surface
- 2) help overcome apical dominance
- 3) kill dicotyledonous weeds in the fields
- 4) speed up the malting process

Ans. 1

109. What amount of energy is released from glucose during lactic acid fermentation?

- 1) More than 18% 2) About 10 %
- 3) Less than 7% 4) Approximately 15%

Ans. 3

110. Read the following statements about the vascular bundles:

- a) In roots, xylem and phloem in a vascular bundle are arranged in an alternate manner along the different radii.
- b) Conjoint closed vascular bundles do not possess cambium
- c) In open vascular bundles, cambium is present in between xylem and phloem
- d) The vascular bundles of dicotyledonous stem possess endarch protoxylem
- (e) In monocotyledonous root, usually there are more than six xylem bundles present

Choose the correct answer from the options given below:

- 1) (b), (c), (d) and (e) only
- 2) (a), (b), (c) and (d) only
- 3) (a), (c), (d) and (e) only
- 4) (a), (b) and (d) only

Ans. NO OPTION

111. Identify the correct set of statements:

- (a) The leaflets are modified into pointed hard thorns in Citrus and Bougainvillea
- (b) Axillary buds form slender and spirally coiled tendrils in cucumber and pumpkin
- (c) Stem is flattened and fleshy in Opuntia and modified to perform the function of leaves
- (d) Rhizophora shows vertically upward growing roots that help to get oxygen for respiration
- (e) Subaerially growing stems in grasses and strawberry help in vegetative propagation

Choose the correct answer from the options given below:

- 1) (a) and (d) only
- 2) (b), (c), (d) and (e) only
- 3) (a), (b), (d) and (e) only
- 4) (b) and (c) only

Ans. 2

112. Which one of the following plants does not show plasticity?

- 1) Coriander 2) Buttercup
- 3) Maize 4) Cotton

Ans. 3

113. Given below are two statements:

Statement I: Cleistogamous flowers are invariably autogamous

Statement II: Cleistogamy is disadvantageous as there is no chance for cross pollination

In the light of the above statements, choose the correct answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but statement II is incorrect
- 3) Statement I is incorrect but statement II is correct
- 4) Both statement I and Statement II are correct

Ans. 4

114. "Girdling Experiment" was performed by Plant Physiologists to identify the plant tissue through which:

- 1) food is transported
- 2) for both water and food transportation
- 3) osmosis is observed
- 4) water is transported

Ans. 1

115. Hydrocolloid carrageen is obtained from:

- 1) Phaeophyceae and Rhodophyceae
- 2) Rhodophyceae only
- 3) Phaeophyceae only
- 4) Chlorophyceae and Phaeophyceae

Ans. 2

116. Given below are two statements:

Statement I: Mendel studied seven pairs contrasting traits in pea plants and proposed the Laws the Inheritance

Statement II: Seven characters examined by Mendel in his experiment on pea plants were seed shape and colour, flower colour, pod shape and colour, flower position and stem height

In the light of the above statement, choose the correct answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but statement II is incorrect
- 3) Statement I is incorrect but statement II is correct
- 4) Both statement I and Statement II are correct

Ans. 4

117. Production of Cucumber has increased manifold in recent years. Application of which of the following phytohormones has resulted in this increased yield as the hormone is known to produce female flowers in the plants:

- 1) Gibberellin
- 2) Ethylene
- 3) Cytokinin
- 4) ABA

Ans. 2

118. The process of translation of mRNA to proteins begins as soon as:

- 1) The larger subunit of ribosome encounters mRNA
- 2) Both the subunits join together to bind with mRNA
- 3) The rRNA is activated and the larger subunit of ribosome encounters mRNA
- 4) The small subunit of ribosome encounters mRNA

Ans. 4

119. Habitat loss and fragmentation, over exploitation, alien species invasion and co-extinction are causes for:

- 1) Competition
- 2) Biodiversity loss
- 3) Nataliy
- 4) Population explosion

Ans. 2

120. Read the following statements and choose the set of correct statement:

- (a) Euchromatin is loosely packed chromatin
- (b) Heterochromatin is transcriptionally active
- (c) Histone octomer is wrapped by negatively charged DNA in nucleosome
- (d) Histones are rich in lysine and arginine
- (e) A typical nucleosome contains 400 bp of DNA helox

Choose the correct answer from the options given below:

- 1) (a), (c), (d) only
- 2) (b), (e) only
- 3) (a), (c), (e) only
- 4) (b), (d), (e) only

Ans. 1

121. Which of the following is incorrectly matched ?

- 1) Ulothrix – Mannitol
- 2) Porphyra – Floridian Starch
- 3) Volvox – Starch
- 4) Ectocarpus – Fucoxanthin

Ans. 1

122. The device which can remove particulate matter present in the exhaust from a thermal power plant is

- 1) Incinerator
- 2) Electrostatic Precipitator
- 3) Catalytic Converter
- 4) STP

Ans. 2

123. DNA polymorphism forms the basis of :

- 1) DNA finger printing
- 2) Both genetic mapping and DNA finger printing
- 3) Translation
- 4) Genetic mapping

Ans. 2

124. Which one of the following statements cannot be connected to Predation ?

- 1) It might lead to extinction of a species
- 2) Both the interacting species are negatively impacted
- 3) It is necessitated by nature to maintain the ecological balance
- 4) It helps in maintaining species diversity in a community

Ans. 2

125. Which one of the following never occurs during mitotic cell division ?

- 1) Movement of centrioles towards opposite poles
- 2) Pairing of homologous chromosomes
- 3) Coiling and condensation of the chromatids
- 4) Spindle fibres attach to kinetochores of chromosomes

Ans. 2

126. Exoskeleton of arthropods is composed of :

- | | |
|----------------|-----------|
| 1) Cellulose | 2) Chitin |
| 3) Glucosamine | 4) Cutin |

Ans. 2

127. What is the net gain of ATP when each molecule of glucose is converted to two molecules of pyruvic acid ?

- | | | | |
|--------|--------|----------|---------|
| 1) Six | 2) Two | 3) Eight | 4) Four |
|--------|--------|----------|---------|

Ans. 2

128. Which one of the following plants shows vexillary aestivation and diadelphous stamens?

- 1) *Pisum sativum*
- 2) *Allium cepa*
- 3) *Solanum nigrum*
- 4) *Colchicum autumnale*

Ans. 1

129. Identify the incorrect statement related to Pollination :

- 1) Pollination by wind is more common amongst abiotic pollination
- 2) Flowers produce foul odours to attract flies and beetles to get pollinated
- 3) Moths and butterflies are the most dominant pollinating agents among insects
- 4) Pollination by water is quite rare in flowering plants

Ans. 3

130. Given below are two statements :

Statement I : Decomposition is a process in which the detritus is degraded into simpler substances by microbes.

Statement II : Decomposition is faster if the detritus is rich in lignin and chitin

In the light of the above statements, choose the correct answer from the options given below :

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II are correct

Ans. 2

131. The appearance of recombination nodules on homologous chromosomes during meiosis characterizes:

- 1) Bivalent
- 2) Sites at which crossing over occurs
- 3) Terminalisation
- 4) Synaptonemal complex

Ans. 2

132. Which of the following produces nitrogen fixing nodules on the roots of *Alnus*?

- | | |
|------------------------|--------------------------|
| 1) <i>Frankia</i> | 2) <i>Rhodospirillum</i> |
| 3) <i>Beijernickia</i> | 4) <i>Rhizobium</i> |

Ans. 1

133. Which one of the following is not true regarding the release of energy during ATP synthesis through chemiosmosis? It involves:
- 1) Breakdown of electron gradient
 - 2) Movement of protons across the membrane to the stroma
 - 3) Reduction of NADP to NADPH₂ on the stroma side of the membrane
 - 4) Breakdown of proton gradient

Ans. 1

134. The flowers are Zygomorphic in:

- a) Mustard
- b) Glumohar
- c) Cassia
- d) Datura
- e) Chilly

Choose the correct answer from the options given below:

- 1) b, c only
- 2) d, e only
- 3) c, d only
- 4) a, b, c only

Ans. 1

135. Match List - I with List II

List - I	List - II
a) Manganese	i) Activates the enzyme catalase
b) Magnesium	ii) Required for pollen germination
c) Boron	iii) Activates enzymes of respiration
d) Iron	iv) Functions in splitting of water during photosynthesis

Choose the correct answer from the options given below:

- 1) a - iv; b - iii; c - ii; d - i
- 2) a - iv; b - i; c - ii; d - iii
- 3) a - iii; b - i; c - ii; d - iv
- 4) a - iii; b - iv; c - i; d - ii

Ans. 1

BOTANY - SECTION - B

136. Addition of more solutes in a given solution will:

- 1) Lower its water potential
- 2) make its water potential zero
- 3) not affect the water potential at all
- 4) raise its water potential

Ans. 1

137. Read the following statements on lipids and find out correct set of statements:

- a) Lecithin found in the plasma membrane is a glycolipid
- b) Saturated fatty acids possess one or more C = C bonds
- c) Gingly oil has lower melting point, hence remains as oil in winter
- d) Lipids are generally insoluble in water but remains as oil in winter
- e) When fatty acid is esterified with glycerol, monoglycerides are formed

Choose the correct answer from the options given below:

- 1) a, d and e only
- 2) c, d and e only
- 3) a, b and d only
- 4) a, b and c only

Ans. 2

138. What is the role of large bundles sheath cells found around the vascular bundles in C₄ plants?

- 1) To increase the number of chloroplasts for the operation of Clavin cycle
- 2) To enable the plant to tolerate high temperature
- 3) To protect the vascular tissue from high light intensity
- 4) To provide the site for photo respiratory pathway

Ans. 1

139. The genetic fleet of buses in Delhi were converted to CNG from diesel. In reference to this, which one of the following statements is false?

- 1) The same diesel engine is used in CNG buses making the cost of conversion low
- 2) It is cheaper than diesel
- 3) It can not be adulterated like diesel
- 4) CNG burns more efficiently than diesel

Ans. 1

140. Transpose can be used during which one of the following?

- 1) Gene silencing
- 2) Autoradiography
- 3) Gene sequencing
- 4) Polymerase Chain Reaction

Ans. 1

141. Match List I with List II

List I	List II
a) Metacentric chromosomes	i) Centromere situated close to the end forming one extremely short and one very long arms
b) Acrocentric chromosome	ii) Centromere at the terminal end
c) Sub metacentric	iii) Centromere in the middle forming two equal arms of chromosomes
d) Telocentric chromosomes	iv) Centromere slightly away from the middle forming one shorter arm and one longer arm

Choose the correct answer from the options given below:

- 1) a - i; b - iii; c - ii; d - iv
- 2) a - ii; b - iii; c - iv; d - i
- 3) a - i; b - ii; c - iii; d - iv
- 4) a - iii; b - i; c - iv; d - ii

Ans. 4

142. Match the plant with the kind of life cycle it exhibits:

List I	List II
a) Spirogyra	i) Dominant diploid sporophyte vascular plant, with highly reduced male or female gametophyte
b) Fern	ii) Dominant haploid free living gametophyte
c) Funaria	iii) Dominant diploid sporophyte alternating with reduced gametophyte called prothallus
d) Cycas	iv) Dominant haploid leafy gametophyte alternating with partially dependent multicellular sporophyte

Choose the correct answer from the options given below:

- 1) a - ii; b - iii; c - iv; d - i
- 2) a - iii; b - iv; c - i; d - ii
- 3) a - ii; b - iv; c - i; d - iii
- 4) a - iv; b - i; c - ii; d - iii

Ans. 1

143. Which of the following occurs due to the presence of autosome linked dominant trait?

- 1) Myotonic dystrophy
- 2) Haemophilia
- 3) Thalessemia
- 4) Sickle cell anaemia

Ans. 1

144. The anatomy of springwood shows some peculiar features. Identify the correct set of statements about springwood.

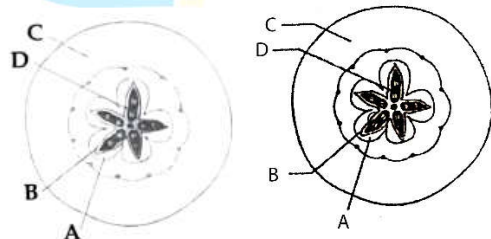
- a) It is also called as the early wood
- b) In spring season cambium produces xylem elements with narrow vessels
- c) It is lighter in colour
- d) The springwood along with autumn wood shows alternate concentric rings forming annual rings
- e) It has lower density

Choose the correct answer from the options given below:

- 1) a, c, d and e only
- 2) a, b and d only
- 3) c, d and e only
- 4) a, b, d and e only

Ans. 1

145. Which part of the fruit, labelled in the given figure makes it a false fruit?



- 1) B - Endocarp
- 2) C - Thalamus
- 3) D - Seed
- 4) A - Mesocarp

Ans. 2

146. In the following palindromic base sequences of DNA, which one can be cut easily by particular restriction enzyme?

- 1) 5'GATTC3' ; 3'CTTAAG5'
- 2) 5'CTCAGT3';3'GAGTCA5'
- 3) 5'GTATTC3';3'CATAAG5'
- 4) 5'GATACT3';3'CTATGA5'

Ans. 1

ZOOLOGY - SECTION - A

147. While explaining interspecific interaction of population, (+) sign is assigned for beneficial interaction, (-) sign is assigned for detrimental interaction and (0) for neutral interaction. Which of the following interactions can be assigned (+) for one species (-) for another species involved in the interaction?

- 1) Amensalism 2) Commensalism
- 3) Competition 4) Predation

Ans. 4

148. Which one of the following will accelerate phosphorous cycle?

- 1) Volcanic activity
- 2) Weathering of rocks
- 3) Rain fall and storms
- 4) Burning of fossil fuels

Ans. 2

149. Given below are two statements: One is labelled as Assertion (A) and the other is labelled as Reason (R)

Assertion (A) Mendel's law of Independent assortment does not hold good for the genes that are located closely on the same chromosomes.

Reason (R): Closely located genes assort independently.

In the light of the above statements, choose the correct answer from the given below:

- 1) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- 2) (A) is correct but (R) is not correct
- 3) (A) is not correct but (R) is correct
- 4) Both (A) and (R) are correct and (R) is the correct explanation of (A)

Ans. 2

150. If a genetic uses the blind approach for sequencing the whole genome of an organism, followed by assignment of function to different segments, the methodology adopted by him is called as:

- 1) Gene mapping
- 2) Expressed sequence tags
- 3) Bioinformatics
- 4) Sequence annotation

Ans. 4

151. Given below are two statements:

Statement I: Autoimmune disorder is a condition where body defense mechanism recognizes its own cells as foreign bodies

Statement II: Rheumatoid arthritis is a condition where body does not attack self-cells
In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II is are correct

Ans. 2

152. Given below are two Statements:

Statement I: The coagulum is formed of network of threads called thrombins

Statement II: Spleen is the graveyard of erythrocytes

In the light of the above Statements, Choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II is are correct

Ans. 3

153. Identify the asexual reproductive structure associated with Penicillium:

- 1) Conidia 2) Gemmules
- 3) Buds 4) Zoospores

Ans. 1

154. Given below are two Statements:

Statement I: The release of sperms into the seminiferous tubules is called spermiation.

Statement II: Spermatogenesis is the process of formation of sperms from spermatogonia.

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II is are correct

Ans. 1

155. Under normal physiological conditions in human being every 100ml of oxygenated blood can deliver ____ ml of O_2 to the tissues.

- 1) 5 ml
- 2) 4 ml
- 3) 10 ml
- 4) 2 ml

Ans. 1

156. In situ conservation refers to:

- 1) Conserve only high risk species
- 2) Conserve only endangered species
- 3) Conserve only extinct species
- 4) Protect and conserve the whole ecosystem

Ans. 4

157. Natural selection where more individuals acquire specific character value other than the mean character value, leads to:

- 1) Direction change
- 2) Disruptive change
- 3) Random change
- 4) Stabilising change

Ans. 1

158. Breeding crops with higher levels of vitamins and minerals or higher proteins and healthier fats is called:

- 1) Bio-remediation
- 2) Bio-fortification
- 3) Bio-accumulation
- 4) Bio-mangification

Ans. 2

159. Which of the following is present between the adjacent bones of the vertebral column ?

- 1) Cartilage
- 2) Areolar tissue
- 3) Smooth muscle
- 4) Intercalated discs

Ans. 1

160. Nitrogenous waste is excreted in the form of pellet or paste by :

- 1) Salamandra
- 2) Hippocampus
- 3) Pavo
- 4) Ornithorhynchus

Ans. 3

161. Which of the following statements with respect to Endoplasmic Reticulum is incorrect ?

- 1) SER is devoid of ribosomes
- 2) In prokaryotes only RER are present
- 3) SER are the sites for lipid synthesis
- 4) RER has ribosomes attached to ER

Ans. 2

162. Which of the following statements are true for spermatogenesis but do not hold true for Oogenesis ?

- a) It results in the formation of haploid gametes
- b) Differentiation of gamete occurs after the completion of meiosis
- c) Meiosis occurs continuously in a mitotically dividing stem cell population
- d) It is controlled by the Luteinising hormone (LH) and Follicle Stimulating Hormone (FSH) secreted by the anterior pituitary
- (e) It is initiated at puberty

Choose the most appropriate answer from the options given below :

- 1) b and c only
- 2) b, d and e only
- 3) b, c and e only
- 4) c and e only

Ans. 4

163. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A) : All vertebrates are chordates but all chordates are not vertebrates.

Reason (R) : Notochord is replaced by vertebral column in the adult vertebrates.

In the light of the above statements, choose the most appropriate answer from the options given below :

- 1) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- 2) (A) is correct but (R) is not correct
- 3) (A) is not correct but (R) is correct
- 4) Both (A) and (R) are correct and (R) is the correct explanation of (A)

Ans. 4

164. In which of the following animals, digestive tract has additional chambers like crop and gizzard ?

- 1) Bufo, Balaenoptera, Bangarus
- 2) Catla, Columba, Crocodilus
- 3) Pavo, Psittacula, Corvus
- 4) Corvus, Columba, Chameleon

Ans. 3

165. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A) : Osteoporosis is characterised by decreased bone mass and increased chances of fractures.

Reason (R) : Common cause of osteoporosis is increased levels of estrogen.

In the light of the above statements, choose the most appropriate answer from the options given below :

- 1) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- 2) (A) is correct but (R) is not correct
- 3) (A) is not correct but (R) is correct
- 4) Both (A) and (R) are correct and (R) is the correct explanation of (A)

Ans. 2

166. In gene therapy of Adenosine Deaminase (ADA) deficiency, the patient requires periodic infusion of genetically engineered lymphocytes because :

- 1) Gene isolated from marrow cells producing ADA is introduced into cells at embryonic stages
- 2) Lymphocytes from patient's blood are grown in culture, outside the body.
- 3) Genetically engineered lymphocytes are not immortal cells
- 4) Retroviral vector is introduced into these lymphocytes.

Ans. 3

167. Select the incorrect statement with reference to mitosis :

- 1) Spindle fibres attach to centromere of chromosomes
- 2) Chromosomes decondense at telophase
- 3) Splitting of centromere occurs at anaphase
- 4) All the chromosomes lie at the equator at metaphase.

Ans. 1

168. In the taxonomic categories which hierarchical arrangement in ascending order is correct in case of animals ?

- 1) Kingdom, Class, Phylum, Family, Order, Genus, Species
- 2) Kingdom, Order, Class, Phylum, Family, Genus, Species
- 3) Kingdom, Order, Phylum, Class, Family, Genus, Species
- 4) Kingdom, Phylum, Class, Order, Family, Genus, Species

Ans. 4

169. Given below are two statements.

Statement I : Restriction endonucleases recognize specific sequence to cut DNA known as palindromic nucleotide sequence.

Statement II : Restriction endonucleases cut the DNA strand a little away from the centre of the palindromic site.

In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II are correct

Ans. 4

170. Given below are two statements.

Statement I : Mycoplasma can pass through less than 1 micron filter size.

Statement II : Mycoplasma are bacteria with cell wall.

In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II are correct

Ans. 2

171. Which of the following is not a connective tissue ?

- 1) Adipose tissue
- 2) Cartilage
- 3) Neuroglia
- 4) Blood

Ans. 3

172. Lippe's loop is a type of contraceptive used as

- 1) Vault barrier
- 2) Non-Medicated IUD
- 3) Copper releasing IUD
- 4) Cervical barrier

Ans. 2

173. At which stage of life the oogenesis process is initiated ?

- 1) Embryonic development stage
- 2) Birth
- 3) Adult
- 4) Puberty

Ans. 1

174. Identify the microorganism which is responsible for the production of an immunosuppressive molecule cyclosporin A :

- 1) Clostridium butylicum
- 2) Aspergillus niger
- 3) Streptococcus cerevisiae
- 4) Trichoderma polysporum

Ans. 4

175. Given below are two statements.

Statement I : Fatty acids and glycerols cannot be absorbed into the blood.

Statement II : Specialized lymphatic capillaries called lacteals carry chylomicrons into lymphatic vessels and ultimately into the blood.

In the light of the above statements, choose the most appropriate answer from the options given below:

- 1) Both Statement I and Statement II are incorrect
- 2) Statement I is correct but Statement II is incorrect
- 3) Statement I is incorrect but Statement II is correct
- 4) Both Statement I and Statement II are correct

Ans. 4

176. In an E.coli strain i gene gets mutated and its product can not bind the inducer molecule. If growth medium is provided with lactose, what will be the outcome ?

- 1) z, y, a genes will be transcribed
- 2) z, y, a genes will not be translated
- 3) RNA polymerase will bind the promoter region
- 4) Only z gene will get transcribed

Ans. 2

177. Tegmina in cockroach, arises from:

- 1) Mesothorax
- 2) Metathorax
- 3) Prothorax and Mesothorax
- 4) Prothorax

Ans. 1

178. Which of the following is not the function of conducting part of respiratory system ?

- 1) Inhaled air is humidified
- 2) Temperature of inhaled air is brought to body temperature
- 3) Provides surface for diffusion of O₂ and CO₂
- 4) It clears inhaled air from foreign particles

Ans. 3

179. Which of the following is a correct match for disease and its symptoms ?

- 1) Tetany - high Ca²⁺ level causing rapid spasms.
- 2) Myasthenia gravis - Genetic disorder resulting in weakening and paralysis of skeletal muscle
- 3) Muscular dystrophy - An auto immune disorder causing progressive degeneration of skeletal muscle
- 4) Arthritis - Inflamed joints

Ans. 4

180. Regarding Meiosis, which of the statements is incorrect ?

- 1) DNA replication occurs in S phase of Meiosis-II
- 2) Pairing of homologous chromosomes and recombination occurs in Meiosis-I
- 3) Four haploid cells are formed at the end of Meiosis-II
- 4) There are two stages in Meiosis, Meiosis-I and II

Ans. 1

181. Detritivores breakdown detritus into smaller particles. This process is called :
- 1) Fragmentation
 - 2) Humification
 - 3) Decomposition
 - 4) Catabolism

Ans. 1

182. If the length of a DNA molecule is 1.1 metres, what will be the approximate number of base pairs ?
- 1) 6.6×10^9 bp
 - 2) 3.3×10^6 bp
 - 3) 6.6×10^6 bp
 - 4) 3.3×10^9 bp

Ans. 4

183. Which of the following functions is not performed by secretions from salivary glands ?
- 1) Digestion of complex carbohydrates
 - 2) Lubrication of oral cavity
 - 3) Digestion of disaccharides
 - 4) Control bacterial population in mouth

Ans. 3

184. A dehydration reaction links two glucose molecules to produce maltose. If the formula for glucose is $C_6H_{12}O_6$ then what is the formula for maltose ?
- 1) $C_{12}H_{22}O_{12}$
 - 2) $C_{12}H_{22}O_{11}$
 - 3) $C_{12}H_{24}O_{11}$
 - 4) $C_{12}H_{10}O_{10}$

Ans. 2

185. If '8' Drosophila in a laboratory population of '80' died during a week, the death rate in the population is _____ individuals per Drosophila per week
- 1) 10
 - 2) 1.0
 - 3) zero
 - 4) 0.1

Ans. 4

ZOOLOGY – SECTION - B

186. Given below are two statements:
- Statement I :** In a scrubber the exhaust from the thermal plant is passed through the electric wires to charge the dust particles.
- Statement II:** Particulate matter (PM 2.5) can not be removed by scrubber but can be removed by an electrostatic precipitator.
- In the light of the above statements, choose the most appropriate answer from the options given below:
- 1) Both Statement I and Statement II are incorrect
 - 2) Statement I is correct but Statement II is incorrect
 - 3) Statement I is incorrect but Statement II is correct

- 4) Both Statement I and Statement II are correct

Ans. 1

187. Which of the following is a correct statement ?
- 1) Bacteria are exclusively heterotrophic organisms.
 - 2) Slime moulds are saprophytic organisms classified under Kingdom Monera.
 - 3) Mycoplasma have DNA, Ribosome and cell wall
 - 4) Cyanobacteria are a group of autotrophic organisms classified under Kingdom Monera.

Ans. 4

188. Ten E.coli cells with ^{15}N - dsDNA are incubated in medium containing ^{14}N nucleotide. After 60 minutes, how many E.coli cells will have DNA totally free from ^{15}N ?
- 1) 40 cells
 - 2) 60 cells
 - 3) 80 cells
 - 4) 20 cells

Ans. 2

189. Select the incorrect statement with respect to acquired immunity
- 1) Anamnestic response is elicited on subsequent encounters with the same pathogen
 - 2) Anamnestic response is due to memory of first encounter
 - 3) Acquired immunity is non-specific type of defense present at the time of birth.
 - 4) Primary response is produced when our body encounters a pathogen for the first time.

Ans. 3

190. Statements related to human Insulin are given below. Which statement(s) is/are correct about genetically engineered Insulin ?
- a) Pro-hormone insulin contain extra stretch of C-peptide
 - b) A-peptide and B-peptide chains of insulin were produced separately in E.coli, extracted and combined by creating disulphide bond between them.
 - c) Insulin used for treating Diabetes was extracted from Catties and Pigs.
 - d) Pro-hormone Insulin needs to be processed for converting into a mature and functional hormone.
 - e) Some patients develop allergic reactions to the foreign insulin.

Choose the most appropriate answer from the options given below :

- 1) (b) only
- 2) (c) and (d) only
- 3) (c), (d) and (e) only
- 4) (a), (b) and (d) only

Ans. 1

191. Which of the following are not the effects of Parathyroid hormone ?

- a) Stimulates the process of bone resorption
- b) Decreases Ca^{2+} level in blood
- c) Reabsorption of Ca^{2+} by renal tubules
- d) Decreases the absorption of Ca^{2+} from digested food
- e) Increases metabolism of carbohydrates

Choose the most appropriate answer from the options given below:

- 1) (b), (d) and (e) only
- 2) (a) and (e) only
- 3) (b) and (c) only
- 4) (a) and (c) only

Ans. 1

192. Select the incorrect statement regarding synapses:

- 1) Electrical current can flow directly from one neuron into the other across the electrical synapse.
- 2) Chemical synapses use neurotransmitters
- 3) Impulse transmission across a chemical synapse is always faster than that across an electrical synapse.
- 4) The membranes of presynaptic and postsynaptic neurons are in close proximity in an electrical synapse.

Ans. 3

193. Match List -I with List -II.

List -I	List-II
a) Bronchioles	(i) Dense Regular Connective tissue
b) Goblet cell	(ii) Loose Connective Tissue
c) Tendons	(iii) Glandular Tissue
d) Adipose Tissue	(iv) Ciliated Epithelium

Choose the correct answer from the options given below :

- 1) a-I, b-ii, c-iii, d-iv 2) a-ii, b-I, c-iv, d-iii
- 3) a-iii, b-iv, c-ii, d-I 4) a-iv, b-iii, c-I, d-ii

Ans. 4

194. The recombination frequency between the genes a & c is 5%, b & c is 15%, b & d is 9%, a & b is 20%, c & d is 24% and a & d is 29%. What will be the sequence of these genes on a linear chromosome ?

- 1) d, b, a, c 2) a, b, c, d
- 3) a, c, b, d 4) a, d, b, c

Ans. 3

195. If a colour blind female marries a man whose mother was also colour blind, what are the chances of her progeny having colour blindness ?

- 1) 50% 2) 75% 3) 100% 4) 25%

Ans. 3

196. Match List-I with List-II.

List-I (Biological Molecules)	List-II (Biological functions)
a) Glycogen	(i) Hormone
b) Globulin	(ii) Biocatalyst
c) Steroids	(iii) Antibody
d) Thrombin	(iv) Storage product

Choose the correct answer from the options given below:

- 1) (a) - (iv), (b) - (ii), (c) - (i), (d) - (iii)
- 2) (a) - (ii), (b) - (iv), (c) - (iii), (d) - (i)
- 3) (a) - (iv), (b) - (iii), (c) - (i), (d) - (ii)
- 4) (a) - (iii), (b) - (ii), (c) - (iv), (d) - (i)

Ans. 3

197. Which of the following is not a desirable feature of a cloning vector ?

- 1) Presence of a marker gene
- 2) Presence of single restriction enzyme site
- 3) Presence of two or more recognition sites
- 4) Presence of origin of replication

Ans. 3

198. Which of the following statements is not true ?

- 1) Sweet potato and potato is an example of analogy
- 2) Homology indicates common ancestry
- 3) Flippers of penguins and dolphins are a pair of homologous organs
- 4) Analogous structures are a result of convergent evolution

Ans. 3

199. Which one of the following statements is correct ?

- 1) The tricuspid and the bicuspid valves open due to the pressure exerted by the simultaneous contraction of the atria
- 2) Blood moves freely from atrium to the ventricle during joint diastole.
- 3) Increased ventricular pressure causes closing of the semilunar valves.
- 4) The atrio-ventricular node (AVN) generates an action potential to stimulate atrial contraction

Ans. 2

200. Match List-I with List-II with respect to methods of Contraception and their respective actions.

List-I	List-II
a) Diaphragms	(i) Inhibit ovulation and Implantation
b) Contraceptive Pills	(ii) Increase phagocytosis of sperm within Uterus
c) Intra Uterine Devices	(iii) Absence of Menstrual cycle and ovulation following Parturition
d) Lactational Amenorrhea	(iv) They cover the cervix blocking the entry of sperms

Choose the correct answer from the options given below:

- 1) (a)-(iv), (b)-(i), (c)-(ii), (d)-(iii)
- 2) (a)-(ii), (b)-(iv), (c)-(i), (d)-(iii)
- 3) (a)-(iii), (b)-(ii), (c)-(i), (d)-(iv)
- 4) (a)-(iv), (b)-(i), (c)-(iii), (d)-(ii)

Ans. 1